

CO₂ Reduction by Combining Methanation With the Blast Furnace



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As a method for remarkably reducing CO₂ emissions from the blast furnace, a method combining carbon recycling technology (methanation) and the blast furnace was proposed, and theoretical examination was carried out by Rist diagram. The results showed that by synthesizing methane from CO₂ in the blast furnace gas and blowing it through the tuyere, CO₂ emissions could be reduced by more than 30% compared with conventional blast furnaces. At that time, it was also clarified that oxygen blast and pre-heating gas blowing from the shaft upper part were effective for the increase of CO₂ reduction.

The 21st Conference of the Parties to the United Nations Framework Convention on Climate Change (COP21), held in November 2015, adopted the Paris Agreement on emissions of greenhouse gases (GHG), beginning with CO₂.¹ Steel works that use mainly the blast furnace process produce large amounts of steel by consuming coal as a primary energy source. In 2019, the Japanese steel industry emitted 150 million tons of CO₂, or approximately 14% of the country's total annual CO₂ emissions of 1.1 billion tons (actual results for 2019).² Thus, reducing CO₂ emissions is an urgent challenge for the steel industry as a CO₂ emission-intensive industry. Although various energy-saving technologies for reducing CO₂ emissions have been developed to date, including burden distribution control techniques and small coke mixed charging, CO₂ emission reduction by operational technologies is approaching its limit.^{3,4} With this background, moves to convert energy sources from coal (carbon) to hydrogen, which doesn't discharge CO₂, in order to realize a drastic reduction in CO₂ emissions can be seen all over the world. As examples, a one-tuyere hydrogen injection test at thyssenkrupp's Duisburg No. 9 blast furnace and the COURSE50 project in Japan may be mentioned.^{5,6} In recent

years, carbon capture and utilization (CCU) has attracted attention as a method for effectively utilizing CO₂ by producing products using CO₂ captured from the atmosphere or from industrial exhaust gases as a feedstock. Utilizing captured CO₂ to produce products that were originally made from fossil fuels will contribute to low carbonization, and CO₂ will not be emitted while it is fixed in the product. Although CO₂ emissions from manufacturing process are allowed in CCU, there is currently an imbalance between the demand for products that can be produced using CO₂ emitted by the steel industry and the amount of those emissions. For example, Japan imports an average of about 1.7 million tons of methanol each year but considering the fact that 0.7 ton of methanol can be produced from 1 ton of CO₂, conversion of all the CO₂ emitted by the steel industry to methanol is not realistic, since the supply of methanol would vastly exceed demand.⁷

Therefore, the authors attempted to solve the above-mentioned market problem by reusing a product produced by CCU as a reducing agent in the steel industry; in other words, by "carbon recycling" to the steel manufacturing process. It should be noted that this approach has the advantage of not contributing to CO₂ emissions, as the carbon